

4 The Importance of Methane

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Overview

Concern about the emissions of Greenhouse Gas (GHG) that causes Climate Change focuses primarily on carbon dioxide (CO₂), which is an unavoidable consequence of burning fossil fuels. However, methane (CH₄) is an increasingly important contributor to Climate Change, because :-

- The Global Warming Potential (GWP) used in GHG accounting underestimates the importance of methane, particularly in the short term;
- CH₄ from some industrial sources is more amenable to mitigation than CO₂;
- Concern about methane emissions from melting permafrost in the Arctic is increasing because these emissions have the potential to dominate Climate Change effects.

Global Warming Potential of CH₄

Reporting of greenhouse gas emissions as CH₄ is now generally based on one tonne of CH₄ having a GWP equal to 25 tonnes of CO₂. A CH₄ GWP of 21 was established in the 1998 second assessment report of the International Panel on Climate Change (IPCC AR2). It was based on the best understanding at that time of the science of the relative cumulative climate effects of CO₂ and CH₄ emissions over the subsequent 100 years. Subsequent IPCC reports have increased that number. The choice of a 100-year time horizon was arbitrary.

However, CH₄ molecules emitted to the atmosphere are destroyed by oxidation on much shorter timescale than 100 years. In contrast CO₂ molecules are stable in the atmosphere and persist until they are eventually absorbed by the ocean. This comparative importance of methane and CO₂ was reflected in the fifth IPCC report (IPCC AR5).

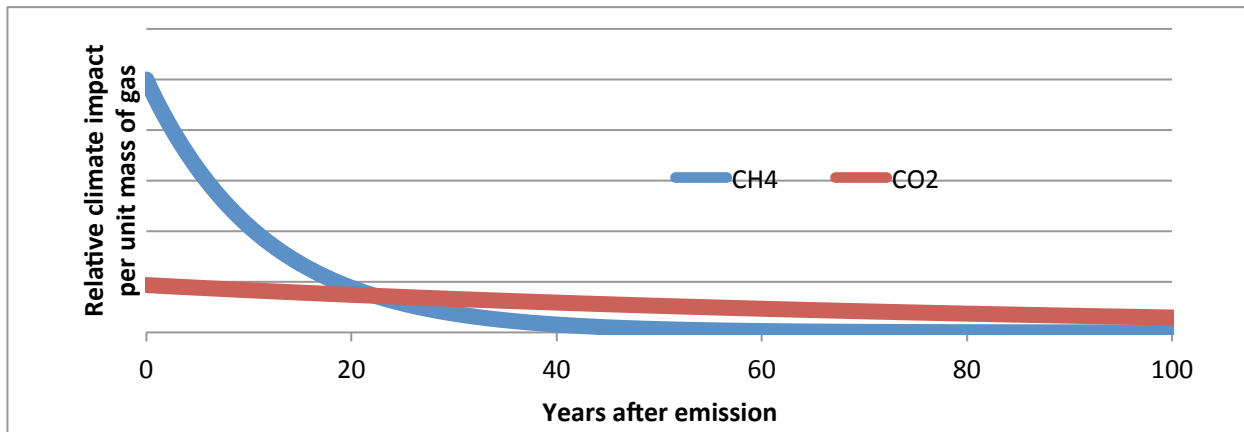
GWP values for fossil CH ₄	100 year time horizon	20 year time horizon
IPCC AR2 (1998)	21	
IPCC AR3 (2003)	23	
IPCC AR4 (2007)	25	72
IPCC AR5 (2013)	30	85

The science underpinning the IPCC reports has moved on over the last 15 years, the chemistry of both CH₄ molecules and CO₂ molecules in the atmosphere is complicated and is still subject to some uncertainty, but it is now understood by the IPCC scientists that, over a 20 year time frame, the impact of CH₄ is four times more significant than the 1998 basis, which is still used for international reporting and emissions trading.

The following chart is derived from detailed analysis presented in the final draft of Chapter 8 of the fifth assessment report by Working Group 1 of the International Panel on Climate



Change (IPCC- AR5, 2013). It shows that the climate impact of CH₄ is much more significant over the first 20 years after discharge to air than over subsequent years. The IPCC AR5 report shows that the consequential integrated temperature rise effects peak about 10 year after the release of a pulse of CH₄ into the atmosphere. It can therefore be argued that considering the climate impacts of CH₄ over a 20 year time horizon is more meaningful than consideration over a 100 year time horizon.



Pollution control

CO₂ emissions are an inevitable consequence of burning hydrocarbon fuels to release energy. In contrast, some CH₄ emissions are an undesirable side effect of industrial activities and can sometimes be reduced by making the activity more efficient.

The most cost effective way of controlling CH₄ emissions to air is to capture the emitted gas and use it as a fuel, whereby the CH₄ is converted to the much less greenhouse-intensive CO₂. If there is no local application for the use of CH₄ as a fuel, then burning unusable CH₄ in a flare is an effective way of reducing its greenhouse impact.

CH₄ is potentially explosive within the range of 5% to 15% in air. To avoid hazard, CH₄ capture systems for burning or flaring must keep the gas concentration well in excess of 15% in air. If that is not practical then dilution and dispersion of CH₄ is used to ensure that the concentration of the gas in air is kept well below 5%, making combustion or flaring impractical.

Sources of methane

Ruminant animals eat vegetation that is decomposed in their first stomach where methanogenic bacteria produce CH₄, which is breathed out by the animal. Choice of feed sources can reduce the CH₄ production in farm animals and also increase the yield of meat/milk. About 80% to 90% of New Zealand's CH₄ emission inventory is attributed to cattle and sheep. Research into reduction of methane emission from ruminants has shown that CH₄ production can be reduced with a strictly controlled diet in a feed lot. However, open paddock grazing as practiced in New Zealand is less amenable to strict diet control.

Rice paddies and some other agricultural practices also generate CH₄ from anaerobic fermentation underwater in wetlands. This CH₄ source is also known as "marsh gas".

Landfill gas is generated when organic materials rot in the absence of air. Methanogenic bacteria convert the carbon in biodegradable materials to 50:50 CO₂ and CH₄. Some landfill

gas can be captured and used or flared. A similar quantity of CH₄ is generated by waste water treatment.

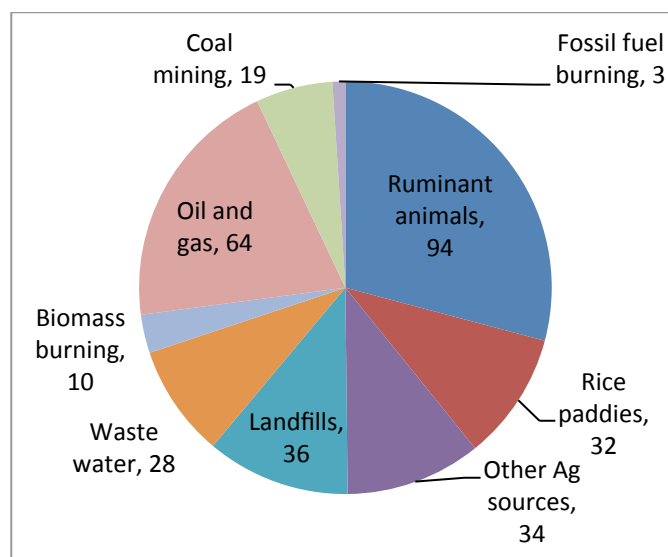
Biomass burning - Open burning of biomass is a significant source of CH₄. The CH₄/CO₂ mole ratio is typically 0.5% to 1%. Hence deforestation and agricultural waste combustion have a secondary greenhouse gas impact. In contrast, the CH₄/CO₂ mole ratio from fossil fuel combustion is much lower.

Oil and Gas production - CH₄ emissions arise from different stages of the oil and gas production, with or without fracking, and also from gas processing and delivery. The oil and gas industry claim that best practice can almost eliminate methane emissions from specific well completion activities. The US natural gas inventory reports average CH₄ losses from well sites as less than 1% of gas produced. The GMI inventory (see below) for the global oil and gas industry, including processing and distribution losses, is equivalent to 2.7% of global natural gas production.

However, CH₄ measurements in air over a gas fields in Colorado have indicated average well site emissions corresponding to 2% - 8% (Pétron et al. 2011). Atmospheric CH₄ measurements in Utah, where environmental controls are less stringent, has indicated well-site emissions at 6% - 11% of production (Karion et al. 2013). Spatial and sectoral analysis of US-wide atmospheric methane data for 2007-8 (Miller et al 2013) determined aggregated methane emissions attributable to oil and gas production in Texas, Oklahoma and Kansas as 2% to 7% of gas production in those years. Gas distribution losses would be additional.

Pre-combustion emissions of CH₄ would eliminate the greenhouse advantage of natural gas over coal for power generation if emissions are greater than about 5% of production, using the AR5 20-year GWP of 85.

Coal mining - CH₄ is adsorbed onto coal underground, typically 0.5% of the energy content of the coal. Deeper coal is usually more gassy. When coal is mined the CH₄ desorbs into the ventilation air in the mine causing a hazard, which has to be managed. Deep unmineable coal seams are sometimes drilled and hydraulically fractured (fracked) as a source of Coal Seam Methane, distributed as natural gas.



Global anthropogenic CH₄ emissions (Mt CH₄) - GMI

The Global Methane Initiative (GMI) has compiled a global inventory of anthropogenic sources of CH₄ emissions (GMI – 2010). The total global CH₄ emissions reported in the GMI inventory for 2010 was 322 million tonnes (Mt) per year, distributed between sources as shown in the above chart. Natural CH₄ releases from wetlands, termites and geologic sources are estimated by GMI to be similar to anthropogenic CH₄ emissions. So the total annual CH₄ emission is over 600 Mt per year.

Using the AR5 20-year GWP of 85, 600 Mt per year would be equivalent to 51 Gigatonnes (Gt) of CO₂ per year. The comparable global emission of CO₂ from fossil fuels in 2010 was 34 Gt CO₂ per year (BP 2012). On this basis methane was a more significant greenhouse gas than CO₂ in 2010.

The current atmospheric concentration of CH₄ is about 1874 ppb in the Northern Hemisphere and 1758 ppb in the Southern hemisphere (IPCC 2013), compared with the pre-industrial level of 722 ppb. The CH₄ content of the global atmosphere is about 5,000 Mt, of which about 550 Mt decomposes each year. Since 2006 the CH₄ content of the atmosphere has been increasing by about 0.5% per year; similar to CO₂.

Methane from the Arctic

The above CH₄ sources are small compared with the potential release of frozen CH₄ from beneath the Arctic Ocean. A recent paper in Nature (Nature 2013) reports *“As the amount of Arctic sea ice declines at an unprecedented rate, the thawing of off-shore permafrost releases methane. A 50-gigatonne (GT) reservoir of methane stored in the form of hydrates, exists on the East Siberian Arctic Shelf. It is likely to be emitted as the seabed warms, either steadily over 50 years or suddenly.”*

A group of concerned scientists has formed the Arctic Methane Warning Group (AMEG 2013) with a mission to warn the world of the potential for the Arctic region to cause global runaway climate change. In contrast a recent wide-ranging US National Academy of Sciences (NAS 2013) report on abrupt climate change considered Russian reports on arctic methane leaks, and concluded it was not an immediate significant hazard. However, those Russian reports are not up to date.

The very large additional source of sub-ocean CH₄ has only recently been quantified. The cut off date for new information for IPCC-AR5 occurred before this phenomenon had been adequately observed and quantified. Therefore its effect on the global CH₄ inventory and the climate change consequences of the release of large amounts of CH₄ from the Arctic are excluded from the 2013 IPCC AR5 report and the GMI inventory. The instability of CH₄ in the East Siberian Arctic Shelf is unusual. At present there is no other unstable large source of CH₄ identified anywhere else.

If this Arctic CH₄ release occurs steadily over 50 years at 1,000 Mt per year, as suggested in the Nature article, then the contribution by CH₄ to Climate Change would be 2.5 times greater than estimated in the GMI data.

A worst case scenario

The IPCC AR5 report indicates that 4×10^{12} tonnes of additional CO_{2-eq} in the atmosphere is likely to result in a 2°C higher global average temperature. A rapid 50,000 Mt release of CH₄

in the Arctic, at the 20-year GWP of 85, would yield an additional 4×10^{12} tonnes of CO₂-eq. That methane emission event alone would be sufficient to raise the global average temperature by 2°C.

However, the temperature rise could be localised to the Arctic region. The Arctic Circle comprises 4% of the surface of the planet. If the methane warming effect remains largely within the Arctic Circle then the localised temperature rise there could be some tens of degrees Celcius.

Greenland lies mostly within the Arctic Circle. After Antarctica, Greenland has the next largest mass of land-based ice in the world. The ice on Greenland is sufficient to raise global sea level by 7 metres, if it were all to slide into the sea.

A worst case scenario is that localised runaway warming occurs in the Arctic due to the feedback effect of rising temperatures releasing more frozen methane causing more regional temperature rise. Then the regional temperature rise causes the ice cap on Greenland to disintegrate and slide into the sea raising global sea level by up to seven metres. Could that worst case scenario play out in the Arctic Region within a decade or two? Nobody knows!

Suggested student assignment

Identify a source of methane emission (e.g. a gas leak) to the atmosphere that can potentially be avoided. Estimate the cost of a repair to avoid that methane emission. Calculate the climate change impact of that CH₄ emission in terms of CO₂ equivalents. Compare the cost of methane emission avoided per tonne of CO₂-equivalent with the current carbon trading price.

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This paper was peer reviewed by Dr Dave Lowe, Co-ordinator, New Zealand – Germany Science + Innovation Relationship; and Peter Barrett, Emeritus Professor of Geology, Antarctic Research Centre, Victoria University of Wellington.

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